Docket No.: 3493-0170PUS1 (PATENT)

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of: Elie LEVERD et al.

Application No.: 10/584,445

Confirmation No.: 4148

Filed: June 22, 2006

Art Unit: 1614

For: PHARMACEUTICAL COPOSITION OF VINFLUNINE WHICH IS INTENDED FOR PARENTERAL ADMINISTRATION

PREPARATION METHOD THEREOF AND USE OF SAME

Examiner: T. P. Thomas

DECLARATION UNDER 37 CFR 1.132

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Madam:

I, Elie Leverd, do declare and say as follows:

- I am one of the Inventors of the above-identified application. A copy of my Curriculum Vitae (five pages) is attached showing my qualifications and experience in the area of pharmaseutical formulations.
- 2. If have reviewed the most recent Office Action dated January 26, 2009 which issued by the US Patent and Trademark Office in connection with the above-identified application. I note that this Office Action includes rejections against the present patent claims based on the combination of two or more of the disclosures of: [1] GlaxoSmithKline ("Prescribing Information; Navelbine (vinorelbine tartrate) Injection", 2002, Nov., pp. 1-17); [2] Duflos '377 (US 6,127,377); [3] Wolgemuth '643 (CA 2,001,643) and [4] Howell ("Anti-vascular effects of

Birch, Stowart, Kolasch & Birch, LLP

ADM//mag

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vinflumine...," British Journal of Cancer (2001) 84 (2), pp. 290-295.). I offer the following expert opinion relevant to the distinctions between the presently claimed and disclosed invention of the above-identified application and these references cited against the present application.

Evidence of Unexpected, Advantageous Properties of Present Invention

- 3. The disclosed and presently claimed invention is directed to a vinflunine pharmaceutical composition in the form of a stable and sterile aqueous solution of a water-soluble vinflunine sait at a pH between 3 and 4, which does not contain any sugar, sugar-based polyol or other preservatives, as recited in claim 1. As explained at pages I-4 of the present specification, conventional pharmaceutical formulations containing vinflunine did not exhibit acceptable storage stability properties or required somewhat complex methods for preparing injectable formulations.
- 4. The improved stability properties exhibited by the composition of the present invention are evidenced by the test results described in connection with Examples 1 and 2 at pages 8-13 of the present specification. These test results establish that the composition of the present invention exhibits significant, unexpected and advantageously improved storage stability properties without requiring complicated techniques or the presence of one or more preservatives. In this regard, note that Example 1 at pages 8-9 of the present application establishes that vinflunine ditartrate is unexpectedly and advantageously more stable in the form of an aqueous solution according to the present invention, as compared to the "pulverulent" form. Example 2 establishes that the aqueous solution according to the present invention ("unbuffered") exhibited unexpected, advantageously improved stability over several formulations containing different buffers at different pH levels. All of the above-noted references cited in the above Office Action fail to disclose or reasonably suggest that these improved stability properties can be obtained by the aqueous formulation of the present invention.

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5. The above-noted advantageous properties exhibited by the aqueous formulation according to the present invention were not conventionally recognized. This is evidenced by the presence of pH buffers and sugar derivatives such as mannitol which are present in the commercialized formulations for vincristine sulfate and vindesine sulfate injection formulations as evidenced by enclosed Exhibit A (Product tabel from http://www.accessdata.fda.gov/scripts/cder/onctools/labels.cfm?GN=vincristine) and Exhibit B (Eldisine@Injection from RxMed: Pharmaceutical information-Eldisine https://www.accessdata.fda.gov/scripts/cder/onctools/labels.cfm?GN=vincristine) and Exhibit B (Eldisine@Injection from RxMed: Pharmaceutical information-Eldisine https://www.accessdata.fda.gov/scripts/cder/onctools/labels.cfm?GN=vincristine) and Exhibit B (Eldisine@Injection from RxMed: Pharmaceutical information-Eldisine https://www.accessdata.fda.gov/scripts/cder/onctools/labels.cfm?GN=vincristine)

Significant Distinctions Between Vinflunine and Vinorelbine

- 6. As disclosed in Duflos '377, vinflunine is structurally related to vinorelbine, but differs in structure since it includes two fluoro substiments and a saturated bond in one of the heterocyclic rings, such that vinflunine is alternatively named as: 20',20'-diffuoro-3',4'-dihydrovinorelbine. Even though vinflunine and vinorelbine may have similar therapeutic properties, these compounds exhibit significantly different physico-chemical properties in at least three areas when in the form of a powder or in the form of an aqueous solution.
- First, there are significant differences in water solubility: vinorelbine tartrate has a solubility higher than 1000 mg/ml whereas vinflunine tartrate has a solubility equal to only 290 mg/ml.
- 8. Second, there are significantly different properties exhibited by each in the form of a powder after 6 months of storage at 5°C and 25°C: vinorelbine tartrate degrades such that the major impurity is due to the oxidation of the alcohol group in the vindeline structure, whereas in contrast, vinflumine ditartrate degrades such that the major impurity is 23-0 dimethylvinflumine which is due to the hydrolysis of the ester group of the vindoline structure. Therefore, vinorelbine tartrate and vinflumine ditartrate generate very different major impurities. This is evidenced by enclosed Exhibits C-1 and C-2 which show that vinorelbine tartrate generates the

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impurity "S/D6" at 5°C and 25°C over time. Exhibit C-I describes the impurity "S/D6" and corresponds to a portion of the documents submitted by the Assignee to appropriate regulatory authorities in order to obtain marketing authorization for the product described in the above-identified application. Exhibit C-2 includes two graphs showing the S/D6 content changes at "ambient" temperature (i.e. 25°C) and at "fridge" temperature (i.e. 5 °C). Labels "521", "522" and "524 correspond to different batches which were tested.

- 9. Third, the process for manufacturing vinflunine is totally different from the process for manufacturing vinorelbine. It is a more complex process since it requires a super acid medium. The differences between these two synthetic routes are illustrated in the enclosed Exhibit D.
- 10. Fourth, vinorelbine exhibits fungicidal activity after up to 28 days of contact with mold spores, with a slight fungicidal activity after 24 hours of contact. In contrast, vinflunine exhibits no fungicidal activity.
- 11. Consequently, several significant properties differ between these two compounds, such that one skilled in the art would not conclude it would be predictable to employ one compound in place of another and expect the same physico-chemical properties to be exhibited together with any improved storage stability properties.
- 12. Vinflunine generally appears to be less stable than vinorelbine since it has a lower solubility. Vinflunine degrades to form a major impurity significantly different from vinorelbine and does not exhibit fungicidal activity as does vinorelbine. Therefore, the behavior and stability of vinflunine in an aqueous solution can not be predicted based on the different physico-chemical properties exhibited by vinorelbine.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are

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punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

June 26, 2009

Date

Enclosures: Curriculum Vitae

Exhibits A, B, C-1, C-2 and D

CURRICULUM VITAE

Last name: LEVERD First name: Elie

Address: 20, Chemin de Cazers-Bas LAMBERT 81100 CASTRES (France)

Date of birth: 6 december 1948
Place of birth: VERSAILLES (France)
Family status; Married - 2 children

QUALIFICATIONS

Appointed Expert Analyst by decree of 21.02.85 published in the Official Bulletin of the Ministry of Social Affairs and National Solidarity N° 85/12 dated 17.04.85

Chemical Engineer – Diploma from the Higher National School of Chemistry – Montpellier 1971

Baccalaureat in elementary mathematics - 1966

CURRENT FONCTION

Head of Liquid Semi-solid and Parenteral Dosage Forms at the PIERRE FABRE DEVELOPMENT CENTER since 1991

PROFESSIONAL EXPERIENCE

Controller of the Department of Formulation Research and Development for Liquid and Semi-solid Dosage Forms at the PIERRE FABRE RESEARCH CENTER from 1979 to 1991

Responsible for Research in the Formulational Department at the PIERRE FABRE RESEARCH CENTER from 1977 to 1979

Responsible for Research in the Phytochemistry Department at the PIERRE FABRE RESEARCH CENTER from 1973 to 1977

Deputy to the controller for physico-chemical control in the Production Unit at PIERRE FABRE MEDICAMENT (1972)

PATENTS

- NEW ANTACID AGENT AND PHARMACEUTICAL COMPOSITIONS CONTAINING IT J.C. TROMBE, G. BONEL, G. MONTEL, F. FAURAN, E. LEVERD, F. MAYNE French Patent n° 81 41080
- ANTACID AND ANTI-REFLUX PHARMACEUTICAL COMPOSITION J.C. TROMBE, G. BONEL, G. MONTEL, F. FAURAN, E. LEVERD, F. MAYNE French Patent n° 81.14081
- INHALER RELEASING PRE-DOSED QUANTITIES OF VOLATILE ACTIVE INGREDIENT OR OBTAINABLE BY A VOLATILE EXCIPIENT CONTAINED IN MICROCAPSULES DISTRIBUTED PERIODICALLY ON A SUPPORT AS WELL AS THE SAID SUPPORT AND ITS MANUFACTURING PROCESS
 M. TRAISNEL, A. GAYOT, P. LETERME, F. MAYNE, E. LEVERD French Patent no 81 21383
- TOOTHPASTE COMPOSITION
 E. LEVERD, M. BAUER, F. MAYNE
 French Patent nº 82 17267
- PHARMACEUTICAL COMPOSITIONS WITH A KERATOLYTIC ACTIVITY IN THE FORM OF HYDRO-ALCOHOLIC GEL E. LEVERD, M. LACROUX French Patent nº 85 11796
- STABLE AQUEOUS SOLUTION OF VINCRISTINE SULPHATE E. LEVERD, M. BAUER, S. BASQUIN French Patent n° 86.06030
- PHARMACEUTICAL COMPOSITION FOR THE PARENTERAL ADMINISTRATION OF NAVELBINE
 E. LEVERD, M. BAUER, S. BASQUIN French Patent n° 87.15686
- COMPOSITION AND METHOD OF PREPARATION OF A DESONIDE EMULSION E. LEVERD, M. LACROUX, D. MROZ-COMBESSIS, J. BOUGARET French Patent n° 94.07445
- CUTANEOUS FOAMING PHARMACEUTICAL COMPOSITION FOR THE TREATMENT OF PITYROSPORUM OVALE INDUCTED DERMATOSIS
 E. LEVERD, M. LACROUX, D. MROZ-COMBESSIS, J. BOUGARET, Ch. MROZ French Patent n° 95 13202

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 WO 0001371
- ANILIDE AND CYCLODEXTRIN COMPLEXES, THEIR PREPARATION AND THEIR USE AS MEDICINE IN PARTICULAR FOR TREATING DYSLIPIDEMIAE J. BOUGARET, E. LEVERD, M.D. IBARRA, A.GIL French Patent n° 01.04855 W0 02/083632
- « VINFLUNINE INJECTABLE »
 E. LEVERD, J. BOUGARET, M.D. IBARRA French Patent n° 03.15312
 European Patent
- STABLE SOLID DISPERSION OF A DERIVATIVE OF VINCA ALKALOID AND PROCESS FOR MANUFACTURING IT
 LEVERD, J. BOUGARET, M.D. IBARRA
 French Patent n° 04.14069
 USA 25348
 WO 2006/069938
- FREEZE-DRIED INJECTABLE PHARMACEUTICAL COMBINATION OF SEMISYNTHETIC VINCA ALKALOIDS AND CARBOHYDRATE STABLE AT ROOM TEMPERATURE E. LEVERD, J. BOUGARET, M.D. IBARRA French Patent N°06.56044 USA 60/897,059 WO 2008/080968
- PHARMACEUTICAL COMPOSITION CONTAINING TRIPTOLIDE E. LEVERD, J.F. BOE EP07008915.6
- TRANSDERMAL PHARMACEUTICAL COMPOSITIONS CONTAINING A STEROIDAL HORMONE E. LEVERD, J. BOUGARET French Patent n°0759864
- ORAL PHARMACEUTICAL COMPOSITION FOR SOFT CAPSULES CONTAINING VINORELBINE AND METHOD OF TREATMENT (Applicant: R.P. SCHERER TECHNOLOGIES, INC.)
 E. LEVERD, J. BOUGARET, M.M. BOHN, N. HEINTZ French Patent n°0206886
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 CA 2,388,431
 WO 03/101383

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 E. LEVERD, D. BEZIAT, P. HATINGUAIS
 Bull. Ohim, Farm. 117:27-32. 1978
- DOSAGE DE L'AJMALICINE DANS CATHARANTHUS ROSEUS G. DON
 J. GLEYE, E. LAVERGNE de CERVAL, E. STANISLAS, E. LEVERD, D. BEZIAT, P. HATINGUAIS

Annales Pharmaceutiques Françaises. 37: 217-224, 1979

- SAPONOSIDES DE RUSCUS ACULEATUS L. ETUDE DE LA LIAISON GENINE-GLUCIDE P. HATINGUAIS, E. LEVERD
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- QUALITE DES MATIERES PREMIERES ET MISE EN OEUVRE GALENIQUE E. LEVERD
 S.T.P. PHARMA. 4(4): 327-329, 1988
- ERYFLUID^(R) SOLUTION: UNE GALENIQUE ADAPTEE AU TRAITEMENT DE L'ACNE PAPULOPUSTULEUSE A PREDOMINANCE INFLAMMATOIRE A. CHIRON, E. LEVERD Abstract Dermato. 124: 60-61, 1991
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- GUIDE POUR LA FILTRATION STERILISANTE RAPPORT D'UNE COMMISSION SFSTP S.T.P. PHARMA PRATIQUES 8(1): 5-17, 1998
- DEVELOPMENT OF AN EXPERIMENTAL METHOD FOR MEASURING THE ADHESION OF GELIFIED FORMULATIONS USING TEXTURE ANALYSERS
 V. LOVERA, J. BOUGARET, E. LEVERD, E. GOUTAY, P. MICHAUD, F. RODRIGUEZ and C. ROQUES
 - S.T.P. PHARMA SCIENCES 8(3): 183-187, 1998
- StUDY OF THIXOTROPIC BASES FOR THE FILLING OF HARD CAPSULES
 P. LOMBARDIN, M. SEILLER, E. LEVERD, E. GOUTAY, J. BOUGARET and J.L. GROSSIORD
 S.T.P. PHARMA SCIENCES 10(6): 429-437, 2000
- THIXOTROPIC BASES FOR THE FILLING OF HARD CAPSULES: IN VITRO RELEASE
 P. LOMBARDIN, E. LEVERD, E. GOUTAY, J. BOUGARET, J.L. GROSSIORD and M. SEILLER
 S.T.P. PHARMA SCIENCES 12(2): 139-142, 2002

 EVOLUTION OF THE INTERACTION OF A NEW CHEMICAL ENTITY, EFLUCIMIBE, WITH γ-CYCLODEXTRIN DURING KNEADING PROCESS
 A. GIL, A. CHAMAYOU, E. LEVERD, J. BOUGARET, M. BARON and G. COUARRAZE EUROPEAN JOURNAL OF PHARMACEUTICAL SCIENCES 23: 123-129, 2004 Product Label Page 1 of 9

Exhibit A



U.S. Food and Drug Administration * Center for Drug Evaluation and Research FDA Oncology Tools Product Label Details in Conventional Order for vincristine

Select Prescribe for how someone prescribing a medication such as a physician may view the product label section order. Select Prepare for how someone preparing a medication such as a pharmacist or nurse may view the sections Select Administer for how someone administering a medication such as a nurse or patient may view the sections. Please send any errors, omissions, and comments to Send Comment.

Pres	cribe	Prepare	Administer
	Application		
	Supplement Number	014103	
	Box Warning		
	Boxed Warning	experienced in the admit extremely important tha properly positioned befo surrounding tissues duri- sulfate injection may can occurs, the injection of the vein. Local injection of the vein. Local injection of the vein. Local injection of Ineat an injection of Ineat of leakar minimize disconfort an GIVEN INTRATHECA	on should be administered by individuals inistration of vincristine sulfate injection. It is the intravenous needle or catheter be ore any vincristine is injected. Leakage into gritravenous administration of vincristine use considerable irritation. If extravasation uld be discontinued immediately, and any dose should then be introduced into another hyaluronidase and the application of moderate ge help disperse the drug and are thought to the possibility of cellulitis. FATAL IF LLY. FOR INTRAVENOUS USE ONLY. In for the treatment of patients given liften injection.
	Complete Label		
	Formatted in PDF	ONCOVIN	
	Description		
	Mechanism of Action	common flowering herb Originally known as leu and VCR. The molecula C46H56N4O10 • H2SO Vincristine sulfate is a w methanol, freely soluble ethanol. In 98% ethanol, spectrum with maximas a Injection, USP is a steril available for intravenous mL contains vincristine for injection, q.s. Sulfuri for pH control. The pH of	tion is the salt of an alkaloid obtained from a the periwinkle plant (Vinca rosea Linn). To coristine it has also been referred to as LCR r formula for vincristine sulfate is 4. It has a molecular weight of 923.04. White to off-white powder. It is soluble in in water, but only slightly soluble in 95% vincristine sulfate has an ultraviolet to 221 nm (L+47,100). Vincristine Sulfate e, preservative-free, single use only solution sue in 2 mL (1 mg and 2 mg) vials. Each sulfate, 1 mg; mannitol, 100 mg; and water ic acid or sodium hydroxide have been added of Vincristine Sulfate Injection ranges from r manufacture, the air in the containers is
	Generic Drug Name	vincristine sulfate	
	Manufacturer		
	Manufacturer	Gensia Sicor Pharmaceu	ticals, Inc. Irvine, CA 92618

Exhibit B



Complete info on herbals. vitamins & dietary supplements

Illness information

ELDISINE® INJECTION

Lilly

Vindesine Sulfate

Antineoplastic

Herbal & dietary supplements

Dharmaceutical

Information

Travel health information

About RxMed

Our Medical Advisory Board Action And Clinical Pharmacology: The mode of action of vindesine is not completely understood. Like the other vinca alkaloids, vinblastine sulfate (Velbe) and vincristine sulfate (Oncovin), vindesine causes arrest of cells in metaphase mitosis. In addition, in vitro investigation has demonstrated that vindesine prevents invasion of normal tissue by malignant cells. Comparative studies with these 3 alkaloids, however, demonstrate differences in their effect at the molecular level. Vindesine has been shown to be 3 times more potent than vincristine and nearly 10 times more potent than vinblastine in effecting mitotic arrest in tissue culture studies designed to arrest from 10% to 15% of the cells in mitosis. At dose levels that arrest 40% to 50% of the cells in mitosis, vindesine and vincristine are approximately equipotent, and both have 3 times the potency of vinblastine. In addition, qualitative differences are noted among the 3 alkaloids. At the lower dose, vinblastine produced a predominance of postmetaphase cells in which the midbody and chromosomes were strikingly apparent. In contrast, cells exposed to vincristine displayed a ball-type metaphase with compact chromosomes within a shrunken spindle.

Unlike vinblastine, vindesine produced very few postmetaphase cells. The spindles in cells exposed to vindesine were swollen with dispersed chromosomes, in distinct contrast to the closely packed chromosomes seen with vincristine.

Vindesine has displayed oncolvtic activity in patients who have relapsed while on multiple-agent treatment that included vincristine.

In laboratory animals and man, the biliary system is the major route of excretion of vindesine.

Hematologic Effects: Clinically, temporary leukopenia is an expected effect of vindesine and the level of the leukocyte count is an important guide to therapy with this drug. In general, the larger the dose employed, the more profound and longer-lasting the

Impurity A (or S/D6 or 3',4',7,8-tetradehydro,3,4'-dideoxy,3,6-epoxy,6,7-dihydro,-C'-norvincaleukoblastine)

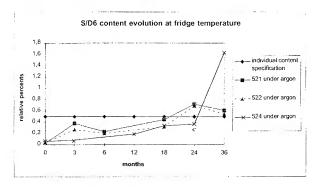
3',4',7,8-tetradehydro,3,4'-dideoxy,3,6-epoxy,6,7-dihydro,-C'-

norvincaleukoblastine, labelled impurity A (or S/D6), comes from the oxidation on the vindoline skeleton of vinorelbine.

It's the major degradation product of raw material.

Extract from DCPB - Part II V6 - March 2000:

The evolution of the major impurity S/D6 at ambient and at fridge temperature is plotted in the 2 graphs below for the various batches. The S/D6 content increases after 3 months refrigerated storage and immediately when the powder is exposed to a temperature of around 25°C.



Related substances (by the liquid chromatography system 2001)

Evolution of major degradation products are summarized below after six months

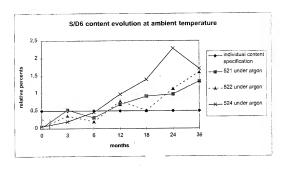
	BATCH 501		BATCH 503	
	25°C 60% RH	5°C	25°C 60% RH	5°C
23.O.demethylvinflunine	+1.1	+0.2	+0.9	+0.1
6'-N-methylvinflunine	+0.7	+0.2	+0.6	+0.1
Δ-7,8-vinflunine-3,6-ether	no	no	+0.3	no
D11(RRT=0.90)	+0.2	no	+0.2	no
4-O-deacetylvinflunine	+0.1	no	+0.2	no
D5(RRT = 0.24)	+0.1	no	+0.1	no
Total content	+2.3	+0.5	+2.3	+0.3

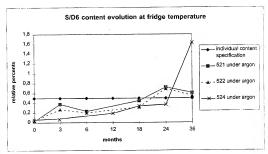
No= no evolution

ILF STABILITY VINORELBINE

Exhibit C-2

The evolution of the major impurity S/D6 at ambient and at fridge temperature is plotted in the 2 graphs below for the various batches. The S/D6 content increases after 3 months refrigerated storage and immediately when the powder is exposed to a temperature of around 25°C.





Ring contraction

crude 1 vinorelbine

Fluorination in super acidic conditions

Purification by chromatography

Ring contraction

crude vinflunine

Exhibit D

Synthesis of vinorelbine tartrate Synthesis of vinflunine ditartrate catharanthine catharanthine hemisulfate hemisulfate catharanthine catharanthine n-oxycatharanthine n-oxycatharanthine vindoline vindoline anhydrovinblastine in dried dichloromethane anhydrovinblastine crude 1 difluoroanhydrovinbla -stine (AVLBF2) crude 2 AVLBF2 purified AVLBF2 isolated AVLBF2

